

## Materials Science Answer Sheet

Team Name: **KEY**

1.  $A = \frac{\pi}{4}d^2$   
 $Creep = A(t) - A_{initial}$  (this is given in problem statement)

2 points for correct area calculations  
3 points for correct force calculations

2. 1 point for correct axes labels and units  
2 points for proper plotting of data from (1)

$$Creep\ Rate = \frac{\Delta Creep}{\Delta Time}$$

1 point for drawing appropriate trendline (or setting up regression equation)  
1 point for correct calculation of creep rate

3. D    4. B    5. C    (2 each)

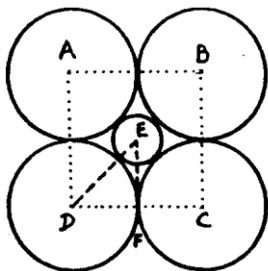
6.  $F = mg$                            $g = 9.81\ m/s^2$   
 $\sigma = \frac{F}{A}$                            $A = \frac{\pi}{4}(0.000511\ m)^2$   
 $\epsilon = \frac{x}{l}$

1 point for correct calculation of force  
2 points for correct calculation of stress  
2 points for correct calculation of strain

7. 1 point for correct axes labels and units  
2 points for proper plotting of data from (6)  
 $E = \frac{\sigma}{\epsilon}$   
1 point for drawing appropriate trendline (or setting up regression equation)  
1 point for calculation of modulus



22.



$\Delta EDF$  is a right-angled triangle. Using Pythagoras theorem,

$$(ED)^2 = (EF)^2 + (DF)^2$$

$$(r+R)^2 = R^2 + R^2$$

$$= 2R^2$$

$$r+R = \sqrt{2}R$$

$$r = (\sqrt{2} - 1)R$$

$$= 0.414R$$

$$\frac{r}{R} = 0.414$$

Hence, the minimum  $r/R$  ratio for C.N. = 6 to be stable is 0.414.

2 points for setting up correct Pythagorean Theorem Equation

2 points for correctly substituting  $r$  and  $R$

1 point for correct answer

23. A

24. B

25. A

26. C

27. A

28. B

29. A

30. D

31. C

(2 each)

32.

$$(\sqrt{2})a = 4r$$

$$a = 4r/\sqrt{2} = 4(1.28 \text{ \AA})/\sqrt{2} = 3.62 \text{ \AA}$$

2 points for identifying correct relationship and equation

2 points for correct substitutions

1 point for correct solution

33.

$$\text{Packing factor} = 4 \left(\frac{4}{3}\right) \pi r^3 / a^3 = 4 \left(\frac{4}{3}\right) \pi r^3 / (4r/\sqrt{2})^3 = 4\left(\frac{4}{3}\right)\pi / [64/(2\sqrt{2})] = 0.74$$

1 point for correctly calculating volume of copper atom

1 point for identifying number of copper atoms in unit cell (4)

1 point for correctly calculating volume of unit cell based on result from 32

1 point for setting up correct packing factor equation

1 point for correct answer

34. {1 1 1}

(1)

35. D

36. B

37. A

38. B

(2 each)

**TIEBREAKERS:**

1. Combined total of 1, 2, 6, 7.
2. Problem 33
3. Problem 22
4. Percent Difference between answer to 7 and known value